

An Efficient Catalytic Stereoselective Route to a Key Intermediate for the Synthesis of the Long-Lived PGI₂ Analog ZK 96480 (Cicaprost™)

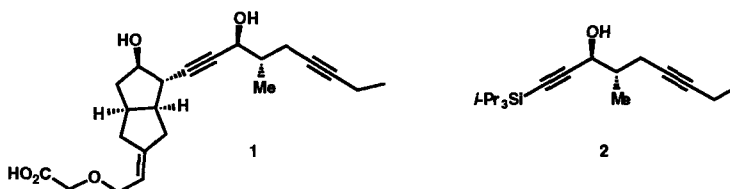
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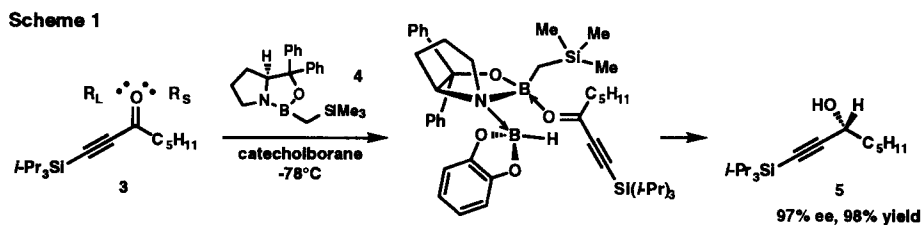
Summary: An expeditious and stereoselective synthesis of a key chiral intermediate (2) for the synthesis of the therapeutically useful PGI₂ analog Cicaprost™ is described.

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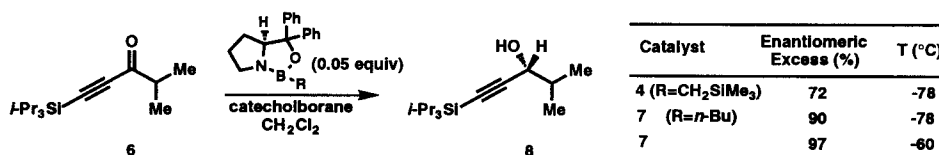
Primary pulmonary hypertension (PPH) is an increasingly common and fatal disease. The only life-sustaining treatments for PPH at present are either 24-hour infusion therapy with prostaglandin I₂ (PGI₂) or combined heart-lung transplant.¹ Long lived, metabolically stable, and orally active PGI₂ analogs such as Cicaprost™ (1)² (dose 0.5 mg/kg, t_{1/2} ca. 1 h), offer the prospect of a far more acceptable dosing regimen for the first option.^{3,4} Unfortunately, published syntheses of Cicaprost are long and possibly impractical.⁵ Described herein is a simple and efficient synthetic route to the key intermediate 2 which corresponds to the omega-sidechain of Cicaprost.



We have recently reported a highly enantioselective catalytic reduction of α,β -ynones to propargylic alcohols as shown in Scheme 1.^{6,7} For example, treatment of ketone 3 and oxazaborolidine 4 (0.05 equiv) with catecholborane (1.2 equiv) in CH₂Cl₂ at -78 °C produced the (*R*)-acetylenic alcohol 5 in 97% enantiomeric excess (*ee*) and 98% yield. The high level of asymmetric induction observed in this process is the result of a repulsive long range steric interaction between the terminal triisopropylsilyl group and the bulky trimethylsilylmethyl substituent on the boron atom of the catalyst, thus favoring binding of the catalyst to the ketone electron lone pair *anti* to the triisopropylsilylalkynyl substituent. As a result, the triisopropylsilylalkynyl substituent functions as the larger group (R_L) versus the *n*-pentyl group (R_S).



The effectiveness of this methodology led us to study the reduction of isopropyl ketone **6** to alcohol **8** as a model for the preparation of **2**. Utilization of (*S*)-B-CH₂SiMe₃ catalyst **4** at -78°C provided **8** in only 72% ee. (*S*)-B-*n*-Bu catalyst **7**, however, resulted in a substantial improvement in enantioselectivity to 90% ee.^{6,8} A further increase to 97% ee was obtained by (1) performing the reduction at -60°C , and (2) pre-cooling the CH₂Cl₂ solution of catecholborane by slow addition down the side of the flask.^{9,10}

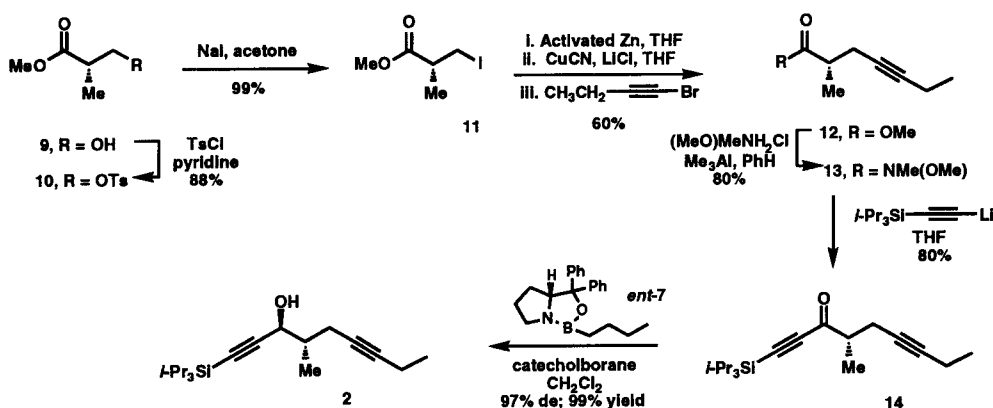


These optimized conditions are ideally suited to the stereoselective synthesis of omega-sidechain **2** as outlined in Scheme 2. Treatment of methyl-(*S*)-(+)-3-hydroxy-2-methylpropionate (**9**)¹¹ and 7 equiv of pyridine at 0°C with 1.3 equiv of *p*-toluenesulfonyl chloride for 10.5 h provided tosylate **10** (88%), which was converted to known iodide **11** under Finkelstein conditions (NaI, acetone, 60°C , 2 h, 99% yield).^{12,13} Iodide **11** and 3 equiv of activated zinc dust¹⁴ in THF were heated at 40°C for 4 h.¹⁵ The resulting organozinc reagent was added to 0.9 equiv of copper(I) cyanide and 1.8 equiv of LiCl in THF at -10°C , stirred 10 min, cooled to -78°C , treated with 1-bromo-1-butyne,¹⁶ and warmed to -60°C for 20 h to provide alkyne ester **12** in 60% yield.^{13,17} *N,O*-Dimethylhydroxylamine hydrochloride (3 equiv) in benzene at 0°C was treated with 3 equiv of Me₃Al (2 M in hexanes), and stirred 1 h.¹⁸ Alkyne ester **12** (1 equiv) was added and the solution was heated at 50°C for 3 h. The reaction was quenched by the addition of MeOH, diluted with CH₂Cl₂, stirred with Na₂SO₄·10H₂O (9 equiv), and filtered to afford after silica gel chromatography Weinreb amide **13** in 80% yield, $[\alpha]_{\text{D}}^{23} +6.9$ (*c* 1.3, CH₂Cl₂). Lithium triisopropylsilylacetylide (1.3 equiv of triisopropylsilylacetylene¹⁹, 1.2 equiv of *n*-BuLi, THF, -78°C to 0°C , 30 min) was added to a solution of **13** in THF at -10°C . After 30 min the reaction was quenched with saturated NH₄Cl solution to give ketone **14** in 80% yield, $[\alpha]_{\text{D}}^{23} -10.2$ (*c* 1.3, CH₂Cl₂). To a solution of **14** and 0.05 equiv of (*R*)-B-*n*-Bu catalyst *ent*-**7** in CH₂Cl₂ at -60°C was added 1.2 equiv of catecholborane in CH₂Cl₂ which afforded, after 3 h, alcohol **2** in 97% diastereomeric excess (de) and 99% yield,

$[\alpha]_D^{23} +40.8$ (c 1.75, CH_2Cl_2).^{20,21} Utilization of catalyst **7** in the reduction of **14** produced the other diastereomer in 96% de and 99% yield, ($[\alpha]_D^{23} -6.4$ (c 0.55, CH_2Cl_2)), demonstrating that the reduction of chiral ketone **14** is a reagent-controlled process.

The stereoselective preparation of **2** described herein proceeds in 6 steps and 33% overall yield starting from readily available methyl-(*S*)-(+)-3-hydroxy-2-methylpropionate (**9**). This synthesis highlights an effective carbon-carbon bond formation utilizing an enantiopure homoenolate for the rapid construction of requisite ketone **14** and a practical application of long range steric effects in the chiral oxazaborolidine-catalyzed reduction of ketones.²²

Scheme 2



References and Notes

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9. The absolute stereochemistry of **8** was determined to be *R* by (1) benzylation (BzCl, 4-(dimethylamino)pyridine (DMAP), Et₃N, CH₂Cl₂, 97% yield), (2) alkyne oxidation to the carboxylic acid (RuCl₃ (cat), NaIO₄, CH₃CN, CCl₄, H₂O, 66% yield), and (3) thioesterification (ClC(O)OEt, Et₃N, CH₂Cl₂, 0 °C; EtSH, DMAP, 74% yield) to provide *S*-ethyl-(*R*)-2-(benzoyloxy)-3-methylthiobutyrate: $[\alpha]_D^{23}$ -17.4 (*c* 1.75, CHCl₃), lit. for (*S*)-isomer $[\alpha]_D^{23}$ +18.9 (*c* 0.94, CHCl₃). See: Falck, J. R.; Bhatt, R. K.; Ye, J. *J. Am. Chem. Soc.* **1995**, *117*, 5973.
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20. Reduction of **14**: Ketone **14** (45 mg, 0.148 mmol, azeotropically dried with toluene) was treated with (*R*)-*B*-*n*-Bu catalyst *ent*-**7** (74 μL, 0.007 mmol, 0.1 M solution in toluene). Toluene was removed *in vacuo*, CH₂Cl₂ (900 μL) was added, the solution was cooled to -60 °C, and catecholborane (23 mg, 20 μL, 0.192 mmol) in CH₂Cl₂ (142 μL) was added down the side of the flask over 10 min. After 3 h, MeOH (200 μL) was added, the solution was warmed to room temperature, diluted with hexanes (3 mL), and applied directly to a silica gel column for purification (hexanes to 10:1 hexanes-Et₂O). This provided 45 mg (>99% yield) of **2** as a clear oil; $[\alpha]_D^{23}$ +40.8 (*c* 1.75, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 4.41 (t, J=5.6 Hz, 1H), 2.28 (m, 2H), 2.16 (qt, J=2.2, 7.5 Hz, 2H), 1.93 (m, 2H), 1.12 (m, 5H), 1.07 (d, J=2.7 Hz, 21H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 107.021, 86.769, 83.409, 66.524, 39.259, 22.324, 18.609, 14.952, 14.311, 12.458, 11.199 ppm; FT-IR (neat) 3400, 2960, 2943, 2893, 2866, 883 cm⁻¹; MS (CI): 324 ([M+NH₄]⁺, 100); HRMS (CI): calcd for [C₁₉H₃₄OSi + NH₄]⁺: 324.2723; found: 324.2736. Conversion of the alcohols to the *p*-nitrobenzoates (*p*-nitrobenzoylchloride, DMAP, CH₂Cl₂, 23 °C) and HPLC analysis (Chiralcel OD, 0.25% *i*-PrOH in hexanes, 0.6 mL/min, λ=235 nm) showed the product to be of 97% de (R_t: 29.8 min, major; 35.6 min, minor).
21. The relative stereochemistry of the reduction of ketone **14** to alcohol **2** was assigned by analogy with the reduction of ketone **6** to alcohol **8**.
22. This research was supported by grants from the National Institutes of Health and the National Science Foundation.

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